

**SEPARATION AND PURIFICATION OF POLY-3-HYDROXYALKANOIC ACID**

Publication number: JP2001046094

Publication date: 2001-02-20

Inventor: ODAWARA OSAMU; MIYAMOTO KENJI; YOKOMIZO SATOSHI; MATSUMOTO KEIJI

Applicant: KANEYAFUCHI CHEMICAL IND

Classification:

- International: C12P7/62; C12P7/62; (IPC1-7): C12P7/62; C12P7/62; C12R1/05

- european:

Application number: JP19990226841 19990810

Priority number(s): JP19990226841 19990810

[Report a data error here](#)

**Abstract of JP2001046094**

**PROBLEM TO BE SOLVED:** To separate and purify the subject compound which is a biodegradable plastic by adding a surfactant to a suspension of a microbial cell of a microorganism containing a poly-3-hydroxyalkanoic acid and carrying out a physical crushing treatment of the resultant mixture liquid.

**SOLUTION:** A surfactant is added to a suspension of a microbial cell of a microorganism (e.g. *Aeromonas caviae*) containing a poly-3-hydroxyalkanoic acid comprising a copolymer of D-3-hydroxybutyrate and D-3-hydroxyhexanoate, a terpolymer, etc., of the D-3-hydroxybutyrate, D-3-hydroxyvalerate and D-3-hydroxyhexanoate and the resultant mixture liquid is then subjected to a physical crushing treatment to thereby separate and purify the objective poly-3-hydroxyalkanoic acid suitable as a raw material, etc. for a plastic product, an implant material without requiring recovery, a drug carrier, a fertilizer carrier, an agricultural mulching film, a fishing gear such as a fishing line, a bag, etc. such as a refuse bag, etc. for composts in high purity and high yield.

---

Data supplied from the [esp@cenet](mailto:esp@cenet) database - Worldwide

(19)日本国特許庁 (JP)

(12) 公開特許公報 (A)

(11)特許出願公開番号

特開2001-46094

(P2001-46094A)

(43)公開日 平成13年2月20日(2001.2.20)

(51)Int.Cl.  
C 12 P 7/62  
// (C 12 P 7/62  
C 12 R 1:05)

識別記号

F I  
C 12 P 7/62

フ-72-1\*(参考)  
4 B 0 6 4

審査請求 未請求 請求項の数3 O.L (全5頁)

(21)出願番号 特願平11-226841

(22)出願日 平成11年8月10日(1999.8.10)

(71)出願人 000000941  
鐘源化学工業株式会社  
大阪府大阪市北区中之島3丁目2番4号  
(72)発明者 小田原 修  
兵庫県高砂市西畠1丁目13番1-303  
(72)発明者 宮本 慶二  
兵庫県明石市別所町12-32メゾン別所201  
(72)発明者 桃浦 雄  
兵庫県神戸市垂水区塩屋町6-31-17三青  
荘  
(74)代理人 100095832  
弁理士 鎌田 芳徳

最終頁に続く

(54)【発明の名称】 ポリ-3-ヒドロキシアルカン酸の分離精製方法

(57)【要約】

【課題】 PHAを含有する微生物菌体から、少ない工程で高純度のPHAを高収率で得ることのできるPHAの分離精製方法を提供すること。

【解決手段】ポリ-3-ヒドロキシアルカン酸を含有する微生物菌体の懸濁液に界面活性剤を添加し、得られる混合液を物理的破砕処理することを特徴とするポリ-3-ヒドロキシアルカン酸の分離精製方法。

れない。例えば、アルカリゲネス・リポリチカ (*A lacaligenes lipolytica*)、アルカリゲネス・ユウトロファス (*A. eutrophus*)、アルカリゲネス・ラタス (*A. latas*) 等のアルカリゲネス属 (*Alcaligenes*)、シュウドモナス属 (*Pseudomonas*)、バチルス属 (*Bacillus*)、アゾトバクター属 (*Azotobacter*)、ノカルディア属 (*Nocardia*)、エアロモナス属 (*Aeromonas*) の菌が挙げられ、中でも、エアロモナス・キャビエ (*Aeromonas caviae*) 等の菌株、またはエアロモナス・キャビエ由来の PHA 合成酵素群の遺伝子が導入された菌株、例えば、アルカリゲネス・ユウトロファス AC 32 (寄託番号 FERM P-15786) (J. Bacteriol., 179, 4821-4830 頁 (1997)) 等がより好ましい。

【0010】これらの微生物の培養方法は、PHA を多量に効率よく菌体内に蓄積できるものであれば特に限定ではなく、例えば、前記アルカリゲネス・ユウトロファス AC 32 (FERM P-15786) を用いる場合には、J. Bacteriol., 179, 4821-4830 頁 (1997) 等に記載の方法が好ましい。

【0011】本発明におけるボリ-3-ヒドロキシアルカン酸 (PHA) とは、D-3-ヒドロキシブレート (3HB) のホモポリマーや 3HB と他の 3-ヒドロキシアルカン酸との共重合体などを示すが、中でも 3HB と D-3-ヒドロキシキサノン (3HH) との 2 成分共重合体 (*Macromolecules*, 28, 4822-4828 (1995)) または 3HB と D-3-ヒドロキシバレート (3HV) と 3HH との 3 成分共重合体 (特開平 8-289797 号公報) が、物性的面からより好ましい。ここで 3HB と 3HH の 2 成分共重合体を構成する各モノマユニットの組成比については、特に限定されるものではないが、3HB の含有量が 1~99 モル%、3HH ユニットの含有量が 1~99 モル% のものが好適である。また、3HB と 3HV と 3HH との 3 成分共重合体を構成する各モノマユニットの組成比については、特に限定されるものではないが、例えば、3HB ユニットの含有量が 1~95 モル%、3HV ユニットの含有量が 1~96 モル%、3HH ユニットの含有量が 1~30 モル% のものが好適である。また、これらの PHA の分子量は、1 00 万以上が好ましく、50 万以上がより好ましい。

【0012】PHA の微生物菌体中の含有率は、高い方が好ましいのは当然であり、工業レベルでの適用においては乾燥菌体中に 20 重量% 以上が好ましく、界面活性剤処理、物理的破砕処理、分離操作、分離した PHA の純度等を考慮すると 50 重量% 以上が特に好ましい。

【0013】本発明においては、前記のように培養して得られた微生物菌体の懸濁液に界面活性剤を添加する。なお、本発明における「微生物菌体の懸濁液」とは、培養終了後の培養懸濁液または培養液から遠心分離等で分

離した菌体を水に懸濁させた水性の懸濁液を意味する。該懸濁液中における菌体濃度は、湿菌体換算で 500 g / l 以下が好ましく、300 g / l 以下がさらに好ましい。

【0014】本発明で使用する界面活性剤としては、陰イオン性、陽イオン性、両性もしくは非イオン性でも良く、具体的には、デシル硫酸ナトリウム、デシルスルホン酸ナトリウム、コール酸ナトリウム、デオキシコール酸ナトリウム、オレイン酸ナトリウム、セチルトリ

メチルアンモニウムブロミド、デシルビリジニウムクロリド、3-((3-コラミドプロピル)ジメチルアンモニオ) -1- ブロパンスルホン酸、3-((コラミドプロピル)ジメチルアンモニオ) -2- ヒドリキシ -1- ブロパンスルホン酸、デシル -N-ベタイン、オクチルグルコシド、ヘプチルチオグルコシド、ポリエチルエチレングリコールデシルエーテル、ポリオキシエチレンソルブタブフェニルエーテル、ポリオキシエチレンノルフィニルエーテル、ポリオキシエチレンソルビトールエスチル等が挙げられるが、これらに制限されるものではない。

20 本発明においては、特にデシル硫酸ナトリウム、デシルスルホン酸ナトリウム、コール酸ナトリウム、デオキシコール酸ナトリウム、オレイン酸ナトリウム等が、価格、使用量や添加効果の点から好ましい。

【0015】界面活性剤の添加量は、特に制限されないが、微生物菌体重量 (湿菌体換算) 100 重量部に対して、0.001~50 重量部が好ましく、1~20 重量部がより好ましい。濃縮添加量は、界面活性剤の添加効果が良好な観点から、0.001 重量部以上が好ましく、低コストである観点から、50 重量部以下が好ましい。

また、得られた混合液は、PHA 以外の菌体構成成分の可溶化を促進させる観点から、室温下で 1 分~2 時間程度搅拌することが好ましい。

【0016】次いで、前記混合液を物理的破砕処理する。本発明においては、かかる物理的破砕処理を行なうことにより、前記微生物菌体を破砕して PHA を菌体外に漏出させる効果を有する。

【0017】本発明における物理的破砕処理とは、超音波による破砕、高圧ホモジナイザーやミル等による破砕等が挙げられる。高圧ホモジナイザーとしては、独国内の A.P.V. - ゴーリン社製「マントンゴーリン (商品名)」、デンマークの A.P.V. ラニー社製「ミニラボ (商品名)」、米国のマイクロフルイディックス (Microfluidics) 社製「マイクロフルイタイザ (商品名)」等が挙げられ、ミルとしては、スイスのウイリー・エー・バッカフオーフェン (Willy A. Bachofen) 社製「ダイノーミル (商品名)」等が挙げられるが、同等の破砕効果が得られるこれらに限定されるものではない。

【0018】物理的破砕処理の条件としては、用いる手段により一概には限定できないが、例えば、超音波による破砕の場合、米国のブランソン (Branson) 社製、ソニ

なわなかつ以外は同様の操作を行なつた。その結果、遠心分離しても沈殿物は得ることはできず、ポリマーは全く分離できなかつた。

【0031】比較例2

実施例1においてドデシル硫酸ナトリウム処理を行なわなかつ以外は同様の操作を行なつた。その結果、遠心分離して得られた沈殿物の純度は、懸濁前のポリ(3HB-c o-3HH)含有微生物菌体の純度と同じ50%であった。

【0032】比較例3

ポリ(3HB-c o-3HH)含有微生物菌体の懸濁液100mlを「ダイノーミル」を用いて1l/hの流速で1時間破断処理した後、10g/lになるようにドデシル硫酸ナトリウムを加えて室温で1時間攪拌した。得られた菌体懸濁液は非常に粘度で、遠心分離処理してもポリ(3HB-c o-3HH)を得ることはできなかつた。

【0033】以上の結果より、実施例1~3で得られた\*

\*ポリ(3HB-c o-3HH)は、いずれも界面活性剤を使用していない比較例2で得られたものに比べ、高純度のものであることがわかる。

【0034】また、実施例1~3及び比較例1~3の結果より、ポリ(3HB-c o-3HH)の分離精製方法には、物理的破砕処理と界面活性剤の添加の両方が必要であるが、界面活性剤の添加処理液を物理的破砕処理することにより顕著な効果が得られる。

【0035】

【発明の効果】本発明によれば、高純度のポリ-ヒドロキシアルカン酸(PHA)を効率よく、極めて簡便に得られるため、本発明は、PHAの工業的生産の効率向上およびコストの低減に大きく寄与するものである。また、本発明により得られるPHAは、実用品として十分に高い純度を有するものであり、例えば、プラスチック製品、回収不要のインプラント材料、柔軟担体、肥料担体、農業用マルチフィルム、釣糸等の漁具、コンポスト用ゴミ袋等の原料として好適に用いられる。

---

フロントページの続き

(72)発明者 松本 圭司  
兵庫県西宮市大森町11-33

Fターム(参考) 4B064 AD83 BA18 CA02 CA19 CC24  
CE02 CE03 DA01 DA16

## PATENT ABSTRACTS OF JAPAN

(11)Publication number : 2001-046094

(43)Date of publication of application : 20.02.2001

(51)Int.Cl.

C12P 7/62  
//C12P 7/62  
C12R 1:05 )

(21)Application number : 11-226841

(71)Applicant : KANEGAFUCHI CHEM IND CO LTD

(22)Date of filing : 10.08.1999

(72)Inventor : ODAWARA OSAMU  
MIYAMOTO KENJI  
YOKOMIZO SATOSHI  
MATSUMOTO KEIJI

## (54) SEPARATION AND PURIFICATION OF POLY-3-HYDROXYALKANOIC ACID

## (57)Abstract:

**PROBLEM TO BE SOLVED:** To separate and purify the subject compound which is a biodegradable plastic by adding a surfactant to a suspension of a microbial cell of a microorganism containing a poly-3-hydroxyalkanoic acid and carrying out a physical crushing treatment of the resultant mixture liquid.

**SOLUTION:** A surfactant is added to a suspension of a microbial cell of a microorganism (e.g. *Aeromonas caviae*) containing a poly-3-hydroxyalkanoic acid comprising a copolymer of D-3-hydroxybutyrate and D-3-hydroxyhexanoate, a terpolymer, etc., of the D-3-hydroxybutyrate, D-3-hydroxyvalerate and D-3-hydroxyhexanoate and the resultant mixture liquid is then subjected to a physical crushing treatment to thereby separate and purify the objective poly-3-hydroxyalkanoic acid suitable as a raw material, etc. for a plastic product, an implant material without requiring recovery, a drug carrier, a fertilizer carrier, an agricultural mulching film, a fishing gear such as a fishing line, a bag, etc. such as a refuse bag, etc. for composts in high purity and high yield.

## LEGAL STATUS

[Date of request for examination] 06.08.2004

[Date of sending the examiner's decision of rejection]

[Kind of final disposal of application other than the examiner's decision of rejection or application converted registration]

[Date of final disposal for application]

[Patent number]

[Date of registration]

[Number of appeal against examiner's decision of rejection]

[Date of requesting appeal against examiner's decision of rejection]

[Date of extinction of right]

## \* NOTICES \*

JPO and NCIPI are not responsible for any damages caused by the use of this translation.

- 1.This document has been translated by computer. So the translation may not reflect the original precisely.
- 2.\*\*\*\* shows the word which can not be translated.
- 3.In the drawings, any words are not translated.

---

CLAIMS

---

[Claim(s)]

[Claim 1] The separation purification approach of the Polly 3-hydroxy alkane acid characterized by adding a surfactant to the suspension of the microorganism biomass containing a Polly 3-hydroxy alkane acid, and carrying out physical crushing processing of the mixed liquor obtained.

[Claim 2] The separation purification approach according to claim 1 that a Polly 3-hydroxy alkane acid is 3 component copolymer of 2 component copolymer of D-3-hydroxy butyrate (3HB) and D-3-hydroxy hexanoate (3HH) or D-3-hydroxy butyrate (3HB), D-3-hydroxyvalerate (3HV), and D-3-hydroxy hexanoate (3HH).

[Claim 3] The separation purification approach according to claim 1 or 2 that the microorganism containing a Polly 3-hydroxy alkane acid is the strain into which the Polly 3-hydroxy alkane acid synthetic enzyme group gene of the Aeromonas KYABIE origin was introduced.

---

[Translation done.]

## \* NOTICES \*

JPO and WIPO are not responsible for any damages caused by the use of this translation.

1. This document has been translated by computer. So the translation may not reflect the original precisely.  
2 cases shows the word which can not be translated.  
3. In the drawing, any words are not translated.

## DETAILED DESCRIPTION

## [Detailed Description of the Invention]

[0001]

[Field of the Invention] This invention relates to the separation/purification approach from the microorganism biomass of a Poly-3-hydroxy alcanoic acid.

[0002]

[Description of the Prior Art] Although current and a plenty wastes are processed by incineration, reclamation, etc., there are troubles, such as the waste of the earth and ground reclamation, the disposal of the incinerated products, the recycling system is in progressing with a rise of the social consciousness to plastic waste recycling. However, as for the actual condition, there is much what remains there being a limitation in a recyclable application, could not respond only by incineration, reclamation, and recycle as a plastic solution art actually, and left, is a nature. Then, after all, it is incorporated into the physical change of materials of a nature, the physical processing is the most effective way, but it is not enough to attract the general public's attention, and it is anxious for the utilization.

[0003] Also in these biodegradable plastics, a Poly-3-hydroxy alcanoic acid (PHA) is called hereinafter. PHA is a biodegradable polymer which is harvested and is accumulated as energy we require to enter into the cycle of materials conversion, and also it is a polymer which is added to the carbon cycle process of a nature, and it is expected that there is almost no adverse effect to an ecosystem, it attracts attention especially. Moreover, also in the medical field, it is thought that the implant material of recovery necessities and the utilization as a drug carrier especially.

[0004] Sold PHA forms the microsomes, is accumulated into the microorganism biomass, and in order to use as these as practice, it needs to carry out separation/purification out of a microorganism biomass. As a known approach of carrying out separation/purification of the PHA from a microorganism biomass, when a *dividuum*, there are an approach of PHA dissolving PHA into a solvent, separating the PHA by a solvent, and then separating the PHA by making biomass constituents other than PHA solidifies, and removing. The latter approach is desirable at the point in the inside that this down stream processing is simpler only easily [separation of PHA].

[0005] However, in the case of obtaining PHA by making biomass constituents other than sold PHA solidifying, and removing → *Dividuum*, 18, and 186-209 Page (1855) were → the method of processing biomass suspension by the sodium hypochlorite, solubilizing biomass constituents other than PHA, and obtaining PHA is indicated. *In fact*, *Although this approach is quite easy as a method*, *since it is necessary to use* *to destroy* *the biomass* *to produce* *a large amount* *of* *high* *concentration* *of* *the* *target* *substance*, *which* *is* *an* *actual* *problem*, *which* *is* *dangerous* *in* *that* *the* *removable* *depolymerize* *of* *PHA* *is* *caused* *or* *obtained* *PHA* *remains*. It is thought that it is not suitable for practical use. Moreover, *JP-A-61-6183B*, biomass structure is destroyed by heating/steaming the microorganism biomass suspension containing PHA above 100 deg Celsius, and the PHA is obtained by adding a PHA decomposing agent to it, adding protease processing, and phospholipid/diacyl ferment processing or hydrogen-peroxide processing, and the method of obtaining PHA is indicated. *In order* *to* *desterilize* *and* *deactivate* *and* *degrade* *the* *microorganism* *biomass*, *and* *in* *order* *to* *obtain* *the* *target* *substance*, *which* *is* *the* *target* *substance* *of* *the* *invention*, *which* *is* *desirable*.

<http://www.ipd.nipdg.go.jp/cgi-bin/tran/web.cgi/eje>

2006/07/20

## JP2001-04694A [DETAILED DESCRIPTION]

3/5 ←→ 5

more is desirable.

[0013] In this invention, a surfactant is added to the suspension of the microorganism biomass which sublimated as mentioned above and was obtained. *In addition*, the suspension of a microorganism biomass in which the weight of the biomass which made up the suspension of the biomass separated from the culture medium or the culture medium after culture termination by centrifugal separation etc. The odd mass concentration in this suspension has 500 or less desirable g/l at wet fungus body conversion, and its 500 or less g/l is still more desirable.

[0014] As a surfactant used by this invention, anion nature, cation nature, both cation, or nonionic are sufficient. Specifically Sodium dioctyl sulfate, sodium dioctyl sulfate sodium, cholic acid sodium, A sodium deoxycholate, sodium oleate, cetyl trimethylbenzimidazolone, Dodecyl pyridinium chloride, a 3-(Chloroacetyl)-2-hydroxypropyl trimethyl ammonium sulfate, A 3-(Chloroacetyl)-2-hydroxypropyl trimethyl ammonium sulfate, a heptyl ether, a decyl ether, a decabutane, octyl glucoside, a heptyl glucoside, although poly ethyl ether diester, polyoxyethylene lauryl ether, polyoxyethylene monohydroxy ether, polyoxyethylene sorbitol ether, etc. are mentioned, it is not restricted to these. Especially in this invention, sodium dioctyl sulfate, sodium dioctyl sulfate sodium, cholic acid sodium, a sodium deoxycholate, sodium oleate, etc. are desirable from the point of a price, the amount used, or the additive effectiveness.

[0015] Although especially the addition of a surfactant is not restricted, 1/50 - 50 weight section is desirable to the microorganism biomass weight (wet fungus body conversion) 100 weight section, and the weight of the biomass which made up the suspension of the biomass which sublimated as mentioned above and was obtained. The weight of the biomass which made up the suspension of the biomass separated from the culture medium or the culture medium after culture termination by centrifugal separation etc. The odd mass concentration in this suspension has 500 or less desirable g/l for the obtained mass lower. It is desirable to stir under a room temperature from a viewpoint which promotes solubilization of biomass constituents other than PHA for 1 minute to about 2 hours.

[0016] Subsequently, physical crushing processing of said mixed liquor is carried out. In this invention, it has the effectiveness of crushing said microorganism biomass and making PHA leakage, and if a biomass which performs well, it is a good method. However, the physical crushing processing of a biomass which is not a biomass which performs well, it is not suitable. [0017] With the physical crushing processing in this invention, crushing by crushing the supersonic wave, the high voltage homogenizer, a mill etc. is mentioned. As a high voltage homogenizer, APV and the "MANTON gauze" by the gauze company of a German country named [name] made in U.S. of Germany, J.P. Lamotte's "microhomogenizer" and U.S. micro wave DISKOPHOR are mentioned. The physical crushing processing of a biomass which is not a biomass which performs well, it is a mill. It is a wheel type OPEN (Willy A.Bachofen) of Switzerland. Although "the die no mill (mill)" by the phrase etc. is mentioned, if the equivalent crushing effectiveness is acquired, it will not be limited to these.

[0018] The physical crushing generally increase as conditions for physical crushing processing with the amount to be used, in crushing by the supersonic wave, it is desirable that it is conducted for 30 minutes at an output 5 and *www* cycle 50% made in U.S. (Branson) and using a SONICRA for example. As crushing by the high voltage homogenizer, the micro-homogenizer by KVP (Lamotte of Denmark) used, and is 500 rpm/min. It is the case that it is a high voltage crushing processing of 1 hour or more, it is a mill. It is desirable that it is a physical crushing processing of 1 hour by the flow rate of 1/lh using the die no mill made from PH-Lab A (Kefen) of Switzerland.

[0019] Milling and physical crushing processing is completed by checking whether in crushing processing field conditions are obtained, after performing *at least* *interval* *interval* alignment processing for example, by 300mm for a little centrifuge tube for 10 minutes.

[0020] Next, centrifugal separation of the processing liquid obtained by carrying out physical

involving this approach by heat treatment, down stream processing has facts, such as that the heat in the following process down stream processing increases, and a mostly complicated thing, further.

[0021] moreover, as an approach of having the process which carries out crushing processing of the microorganism biomass to obtain PHA processing with a surfactant, hydrogen-peroxide processing, and the method of separating PHA is proposed → *www* (Patent Publication No. Heisei 11-251451 [right to] [official report] → in order to use a strong tensile hydrogen peroxide, operation on industrial level is difficult. Moreover, the method of crushing a PHA content microorganism biomass with a high voltage homogenizer and separating PHA as a product group is proposed → *www* (Patent Publication No. Heisei 11-251452 [right to] [official report] → in order to use a high purity of this approach, the microorganism biomass suspension suspension about at least 5 times and high voltage processing is not carried out, and is obtained, the highest yield has about 70 - 85%, and the fault of being low.

[0022] *Problem to be Solved by the Invention* The object of this invention is to offer the separation/purification approach of PHA which can obtain PHA of a high grade from the microorganism biomass containing PHA by high yield by the small routing counter.

[0023] *Problem to be Solved by the Invention* That is, this invention relates to the separation/purification approach of PHA characterized by adding a surfactant to the suspension of the microorganism biomass containing a Poly-3-hydroxy alcanoic acid (PHA), and carrying out physical crushing processing of the mixed liquor obtained.

[0024] *Embodiment of the Invention* The microorganism used for this invention will not be limited especially if it is the microorganism which is accumulating PHA in intracellular. For example, *Alcaligenes RIBOFICHA* (*Alcaligenes* sp. *sp.*), *Alcaligenes*, such as *Alcaligenes eutrophus* (*A. eutrophus*), *Alcaligenes* *sp.* *sp.* *AS-1* (*A. eutrophus* *sp.* *AS-1*), *Alcaligenes* *sp.* *sp.* *AS-2* (*A. eutrophus* *sp.* *AS-2*), *Alcaligenes* *sp.* *sp.* *AS-3* (*A. eutrophus* *sp.* *AS-3*), *Alcaligenes* *sp.* *sp.* *AS-4* (*A. eutrophus* *sp.* *AS-4*), *Alcaligenes* *sp.* *sp.* *AS-5* (*A. eutrophus* *sp.* *AS-5*), *Alcaligenes* *sp.* *sp.* *AS-6* (*A. eutrophus* *sp.* *AS-6*), *Alcaligenes* *sp.* *sp.* *AS-7* (*A. eutrophus* *sp.* *AS-7*), *Alcaligenes* *sp.* *sp.* *AS-8* (*A. eutrophus* *sp.* *AS-8*), *Alcaligenes* *sp.* *sp.* *AS-9* (*A. eutrophus* *sp.* *AS-9*), *Alcaligenes* *sp.* *sp.* *AS-10* (*A. eutrophus* *sp.* *AS-10*), *Alcaligenes* *sp.* *sp.* *AS-11* (*A. eutrophus* *sp.* *AS-11*), *Alcaligenes* *sp.* *sp.* *AS-12* (*A. eutrophus* *sp.* *AS-12*), *Alcaligenes* *sp.* *sp.* *AS-13* (*A. eutrophus* *sp.* *AS-13*), *Alcaligenes* *sp.* *sp.* *AS-14* (*A. eutrophus* *sp.* *AS-14*), *Alcaligenes* *sp.* *sp.* *AS-15* (*A. eutrophus* *sp.* *AS-15*), *Alcaligenes* *sp.* *sp.* *AS-16* (*A. eutrophus* *sp.* *AS-16*), *Alcaligenes* *sp.* *sp.* *AS-17* (*A. eutrophus* *sp.* *AS-17*), *Alcaligenes* *sp.* *sp.* *AS-18* (*A. eutrophus* *sp.* *AS-18*), *Alcaligenes* *sp.* *sp.* *AS-19* (*A. eutrophus* *sp.* *AS-19*), *Alcaligenes* *sp.* *sp.* *AS-20* (*A. eutrophus* *sp.* *AS-20*), *Alcaligenes* *sp.* *sp.* *AS-21* (*A. eutrophus* *sp.* *AS-21*), *Alcaligenes* *sp.* *sp.* *AS-22* (*A. eutrophus* *sp.* *AS-22*), *Alcaligenes* *sp.* *sp.* *AS-23* (*A. eutrophus* *sp.* *AS-23*), *Alcaligenes* *sp.* *sp.* *AS-24* (*A. eutrophus* *sp.* *AS-24*), *Alcaligenes* *sp.* *sp.* *AS-25* (*A. eutrophus* *sp.* *AS-25*), *Alcaligenes* *sp.* *sp.* *AS-26* (*A. eutrophus* *sp.* *AS-26*), *Alcaligenes* *sp.* *sp.* *AS-27* (*A. eutrophus* *sp.* *AS-27*), *Alcaligenes* *sp.* *sp.* *AS-28* (*A. eutrophus* *sp.* *AS-28*), *Alcaligenes* *sp.* *sp.* *AS-29* (*A. eutrophus* *sp.* *AS-29*), *Alcaligenes* *sp.* *sp.* *AS-30* (*A. eutrophus* *sp.* *AS-30*), *Alcaligenes* *sp.* *sp.* *AS-31* (*A. eutrophus* *sp.* *AS-31*), *Alcaligenes* *sp.* *sp.* *AS-32* (*A. eutrophus* *sp.* *AS-32*), *Alcaligenes* *sp.* *sp.* *AS-33* (*A. eutrophus* *sp.* *AS-33*), *Alcaligenes* *sp.* *sp.* *AS-34* (*A. eutrophus* *sp.* *AS-34*), *Alcaligenes* *sp.* *sp.* *AS-35* (*A. eutrophus* *sp.* *AS-35*), *Alcaligenes* *sp.* *sp.* *AS-36* (*A. eutrophus* *sp.* *AS-36*), *Alcaligenes* *sp.* *sp.* *AS-37* (*A. eutrophus* *sp.* *AS-37*), *Alcaligenes* *sp.* *sp.* *AS-38* (*A. eutrophus* *sp.* *AS-38*), *Alcaligenes* *sp.* *sp.* *AS-39* (*A. eutrophus* *sp.* *AS-39*), *Alcaligenes* *sp.* *sp.* *AS-40* (*A. eutrophus* *sp.* *AS-40*), *Alcaligenes* *sp.* *sp.* *AS-41* (*A. eutrophus* *sp.* *AS-41*), *Alcaligenes* *sp.* *sp.* *AS-42* (*A. eutrophus* *sp.* *AS-42*), *Alcaligenes* *sp.* *sp.* *AS-43* (*A. eutrophus* *sp.* *AS-43*), *Alcaligenes* *sp.* *sp.* *AS-44* (*A. eutrophus* *sp.* *AS-44*), *Alcaligenes* *sp.* *sp.* *AS-45* (*A. eutrophus* *sp.* *AS-45*), *Alcaligenes* *sp.* *sp.* *AS-46* (*A. eutrophus* *sp.* *AS-46*), *Alcaligenes* *sp.* *sp.* *AS-47* (*A. eutrophus* *sp.* *AS-47*), *Alcaligenes* *sp.* *sp.* *AS-48* (*A. eutrophus* *sp.* *AS-48*), *Alcaligenes* *sp.* *sp.* *AS-49* (*A. eutrophus* *sp.* *AS-49*), *Alcaligenes* *sp.* *sp.* *AS-50* (*A. eutrophus* *sp.* *AS-50*), *Alcaligenes* *sp.* *sp.* *AS-51* (*A. eutrophus* *sp.* *AS-51*), *Alcaligenes* *sp.* *sp.* *AS-52* (*A. eutrophus* *sp.* *AS-52*), *Alcaligenes* *sp.* *sp.* *AS-53* (*A. eutrophus* *sp.* *AS-53*), *Alcaligenes* *sp.* *sp.* *AS-54* (*A. eutrophus* *sp.* *AS-54*), *Alcaligenes* *sp.* *sp.* *AS-55* (*A. eutrophus* *sp.* *AS-55*), *Alcaligenes* *sp.* *sp.* *AS-56* (*A. eutrophus* *sp.* *AS-56*), *Alcaligenes* *sp.* *sp.* *AS-57* (*A. eutrophus* *sp.* *AS-57*), *Alcaligenes* *sp.* *sp.* *AS-58* (*A. eutrophus* *sp.* *AS-58*), *Alcaligenes* *sp.* *sp.* *AS-59* (*A. eutrophus* *sp.* *AS-59*), *Alcaligenes* *sp.* *sp.* *AS-60* (*A. eutrophus* *sp.* *AS-60*), *Alcaligenes* *sp.* *sp.* *AS-61* (*A. eutrophus* *sp.* *AS-61*), *Alcaligenes* *sp.* *sp.* *AS-62* (*A. eutrophus* *sp.* *AS-62*), *Alcaligenes* *sp.* *sp.* *AS-63* (*A. eutrophus* *sp.* *AS-63*), *Alcaligenes* *sp.* *sp.* *AS-64* (*A. eutrophus* *sp.* *AS-64*), *Alcaligenes* *sp.* *sp.* *AS-65* (*A. eutrophus* *sp.* *AS-65*), *Alcaligenes* *sp.* *sp.* *AS-66* (*A. eutrophus* *sp.* *AS-66*), *Alcaligenes* *sp.* *sp.* *AS-67* (*A. eutrophus* *sp.* *AS-67*), *Alcaligenes* *sp.* *sp.* *AS-68* (*A. eutrophus* *sp.* *AS-68*), *Alcaligenes* *sp.* *sp.* *AS-69* (*A. eutrophus* *sp.* *AS-69*), *Alcaligenes* *sp.* *sp.* *AS-70* (*A. eutrophus* *sp.* *AS-70*), *Alcaligenes* *sp.* *sp.* *AS-71* (*A. eutrophus* *sp.* *AS-71*), *Alcaligenes* *sp.* *sp.* *AS-72* (*A. eutrophus* *sp.* *AS-72*), *Alcaligenes* *sp.* *sp.* *AS-73* (*A. eutrophus* *sp.* *AS-73*), *Alcaligenes* *sp.* *sp.* *AS-74* (*A. eutrophus* *sp.* *AS-74*), *Alcaligenes* *sp.* *sp.* *AS-75* (*A. eutrophus* *sp.* *AS-75*), *Alcaligenes* *sp.* *sp.* *AS-76* (*A. eutrophus* *sp.* *AS-76*), *Alcaligenes* *sp.* *sp.* *AS-77* (*A. eutrophus* *sp.* *AS-77*), *Alcaligenes* *sp.* *sp.* *AS-78* (*A. eutrophus* *sp.* *AS-78*), *Alcaligenes* *sp.* *sp.* *AS-79* (*A. eutrophus* *sp.* *AS-79*), *Alcaligenes* *sp.* *sp.* *AS-80* (*A. eutrophus* *sp.* *AS-80*), *Alcaligenes* *sp.* *sp.* *AS-81* (*A. eutrophus* *sp.* *AS-81*), *Alcaligenes* *sp.* *sp.* *AS-82* (*A. eutrophus* *sp.* *AS-82*), *Alcaligenes* *sp.* *sp.* *AS-83* (*A. eutrophus* *sp.* *AS-83*), *Alcaligenes* *sp.* *sp.* *AS-84* (*A. eutrophus* *sp.* *AS-84*), *Alcaligenes* *sp.* *sp.* *AS-85* (*A. eutrophus* *sp.* *AS-85*), *Alcaligenes* *sp.* *sp.* *AS-86* (*A. eutrophus* *sp.* *AS-86*), *Alcaligenes* *sp.* *sp.* *AS-87* (*A. eutrophus* *sp.* *AS-87*), *Alcaligenes* *sp.* *sp.* *AS-88* (*A. eutrophus* *sp.* *AS-88*), *Alcaligenes* *sp.* *sp.* *AS-89* (*A. eutrophus* *sp.* *AS-89*), *Alcaligenes* *sp.* *sp.* *AS-90* (*A. eutrophus* *sp.* *AS-90*), *Alcaligenes* *sp.* *sp.* *AS-91* (*A. eutrophus* *sp.* *AS-91*), *Alcaligenes* *sp.* *sp.* *AS-92* (*A. eutrophus* *sp.* *AS-92*), *Alcaligenes* *sp.* *sp.* *AS-93* (*A. eutrophus* *sp.* *AS-93*), *Alcaligenes* *sp.* *sp.* *AS-94* (*A. eutrophus* *sp.* *AS-94*), *Alcaligenes* *sp.* *sp.* *AS-95* (*A. eutrophus* *sp.* *AS-95*), *Alcaligenes* *sp.* *sp.* *AS-96* (*A. eutrophus* *sp.* *AS-96*), *Alcaligenes* *sp.* *sp.* *AS-97* (*A. eutrophus* *sp.* *AS-97*), *Alcaligenes* *sp.* *sp.* *AS-98* (*A. eutrophus* *sp.* *AS-98*), *Alcaligenes* *sp.* *sp.* *AS-99* (*A. eutrophus* *sp.* *AS-99*), *Alcaligenes* *sp.* *sp.* *AS-100* (*A. eutrophus* *sp.* *AS-100*), *Alcaligenes* *sp.* *sp.* *AS-101* (*A. eutrophus* *sp.* *AS-101*), *Alcaligenes* *sp.* *sp.* *AS-102* (*A. eutrophus* *sp.* *AS-102*), *Alcaligenes* *sp.* *sp.* *AS-103* (*A. eutrophus* *sp.* *AS-103*), *Alcaligenes* *sp.* *sp.* *AS-104* (*A. eutrophus* *sp.* *AS-104*), *Alcaligenes* *sp.* *sp.* *AS-105* (*A. eutrophus* *sp.* *AS-105*), *Alcaligenes* *sp.* *sp.* *AS-106* (*A. eutrophus* *sp.* *AS-106*), *Alcaligenes* *sp.* *sp.* *AS-107* (*A. eutrophus* *sp.* *AS-107*), *Alcaligenes* *sp.* *sp.* *AS-108* (*A. eutrophus* *sp.* *AS-108*), *Alcaligenes* *sp.* *sp.* *AS-109* (*A. eutrophus* *sp.* *AS-109*), *Alcaligenes* *sp.* *sp.* *AS-110* (*A. eutrophus* *sp.* *AS-110*), *Alcaligenes* *sp.* *sp.* *AS-111* (*A. eutrophus* *sp.* *AS-111*), *Alcaligenes* *sp.* *sp.* *AS-112* (*A. eutrophus* *sp.* *AS-112*), *Alcaligenes* *sp.* *sp.* *AS-113* (*A. eutrophus* *sp.* *AS-113*), *Alcaligenes* *sp.* *sp.* *AS-114* (*A. eutrophus* *sp.* *AS-114*), *Alcaligenes* *sp.* *sp.* *AS-115* (*A. eutrophus* *sp.* *AS-115*), *Alcaligenes* *sp.* *sp.* *AS-116* (*A. eutrophus* *sp.* *AS-116*), *Alcaligenes* *sp.* *sp.* *AS-117* (*A. eutrophus* *sp.* *AS-117*), *Alcaligenes* *sp.* *sp.* *AS-118* (*A. eutrophus* *sp.* *AS-118*), *Alcaligenes* *sp.* *sp.* *AS-119* (*A. eutrophus* *sp.* *AS-119*), *Alcaligenes* *sp.* *sp.* *AS-120* (*A. eutrophus* *sp.* *AS-120*), *Alcaligenes* *sp.* *sp.* *AS-121* (*A. eutrophus* *sp.* *AS-121*), *Alcaligenes* *sp.* *sp.* *AS-122* (*A. eutrophus* *sp.* *AS-122*), *Alcaligenes* *sp.* *sp.* *AS-123* (*A. eutrophus* *sp.* *AS-123*), *Alcaligenes* *sp.* *sp.* *AS-124* (*A. eutrophus* *sp.* *AS-124*), *Alcaligenes* *sp.* *sp.* *AS-125* (*A. eutrophus* *sp.* *AS-125*), *Alcaligenes* *sp.* *sp.* *AS-126* (*A. eutrophus* *sp.* *AS-126*), *Alcaligenes* *sp.* *sp.* *AS-127* (*A. eutrophus* *sp.* *AS-127*), *Alcaligenes* *sp.* *sp.* *AS-128* (*A. eutrophus* *sp.* *AS-128*), *Alcaligenes* *sp.* *sp.* *AS-129* (*A. eutrophus* *sp.* *AS-129*), *Alcaligenes* *sp.* *sp.* *AS-130* (*A. eutrophus* *sp.* *AS-130*), *Alcaligenes* *sp.* *sp.* *AS-131* (*A. eutrophus* *sp.* *AS-131*), *Alcaligenes* *sp.* *sp.* *AS-132</i*

[0021] Same agitation was performed except having changed sodium dodecyl sulfate into sodium lauryl sulfate and sodium in example 3 example 1. After drying this precipitate, when the purity of Port (3) HB-cu-3H0 content was determined, it was 84%.

[0022] Same agitation was performed except having not performed fracture agitation by the "die no m/f" in example of comparison 1 example 1. Consequently, even if it carried out centrifugal separation, precipitate could not be obtained, and the polymer was not able to be separated at all.

[0023] Same agitation was performed except having not performed sodium-dodecyl-sulfate processing in example of comparison 2 example 1. Consequently, the purity of the precipitate obtained by carrying out centrifugal separation was the 55% as the purity of the Port (3) HB-cu-3H0 content microorganisms biomass before suspension.

[0024] After carrying out fracture agitation of the 100ml of the suspension of an example of comparison 3 Port (3) HB-cu-3H0 content microorganisms biomass by the rate of flow of 1 l/h for 1 hour using a "die no m/f", sodium dodecyl sulfate was added and it stirred at the room temperature for 1 hour so that it might become 10 g/l. The obtained biomass suspension was very turbid, and even if it carries out centrifugal separation processing, it was not able to obtain Port (3) HB-cu-3H0.

[0025] The above result shows that it is the thing of a high grade compared with what was obtained in the example 2 of a comparison for which no Port (3) HB-cu-3H0 obtained in the examples 1-3 by using the surfactant.

[0026] Moreover, in the physical crushing processing and a surfactant need both to be added for the separation centrifugation of Port (3) HB-cu-3H0 than the rate of examples 1-3 and the example 1-3 of a comparison, remarkable effectiveness is acquired by carrying out physical crushing processing of the addition processing liquid of a surfactant.

[0027]

[Effect of the Invention] According to this invention, it is efficient, and since it is a device very simple, the invention contributes greatly the Poly-3-hydroxy alcanoic acid (PHAA) of a high purity to the improvement in effectiveness of industrial production of PHAA, and reduction of cost. Moreover, PHA obtained by this invention has purity high enough as daily necessities, and is suitable used as raw materials, such as fishing implements, such as a plastic, an implant material of recovery necessities, a drug carrier, fertilizer support, a mulch for agriculture, and a fishing line, and a garbage bag for sewage.

[Translation done.]